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## INTRODUCTION

Unifloral False Acacia (*Robinia pseudoacacia*) honey (liquid due to the high fructose content, very light colored and flavored) may easily be adulterated with high-fructose corn syrup (HFCS), negatively influencing market growth by damaging consumer confidence [1].



Aquaphotomics considers water as a multi-element system that can be described by its multi-dimensional NIR spectra. Since water's H-bonds are present in most natural samples, this analytical approach, using perturbed water in different environments as a mirror for the rest of the molecules in the sample, can be effectively applied to various fields [2].

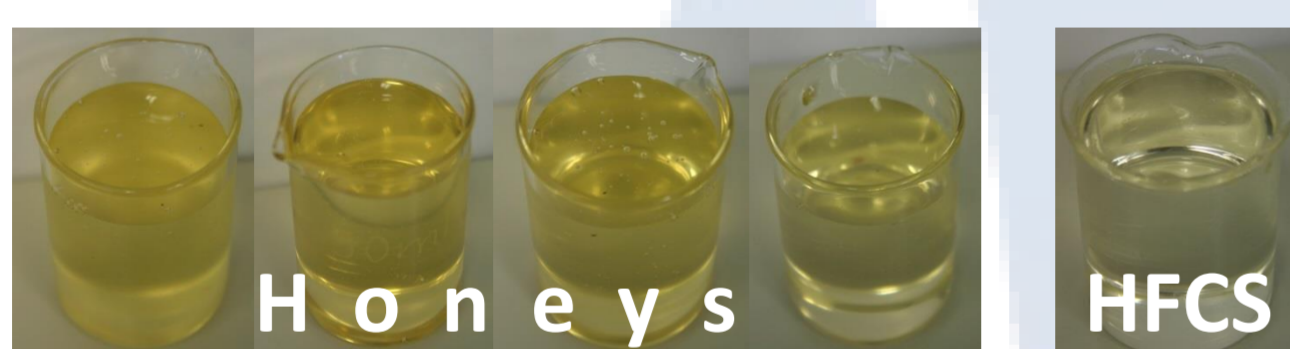
## OBJECTIVE

- Developing an applicable NIR model for screening the adulteration of unifloral *Robinia* honey using fiber-optic probe.
- Applying recent findings of aquaphotomics to interpret the chemometrics calibration models for measuring the level of fructose syrup adulteration.
- Extracting important information about the functionality of *Robinia* honey related to its water structures.

## MATERIALS & METHODS

Pure *Robinia* honey samples

- from four geographic regions of Hungary
- in different periods of False Acacia blossom in 2012.



Isoglucose syrup (High Fructose Corn Syrup, HFCS)

- from high-temperature closed process,
- liquid, cleaned, sterile, ion exchanged and filtered
- 40% fructose, 33% glucose.

Individual honey samples were diluted with HFCS

- random concentrations (n = 40)
- range of honey content = 100-60% (honey content mean±SD = 80.79±12.89)

- FOSS NIRSystems 6500 spectrometer (FOSS NIRSystems, Inc., Laurel, MD, USA)

- OptiProbe fiber optic immersion sampling unit, with 2mm layer thickness

- Transflectance spectra, 1100–1800 nm, 2nm step

- Scanning in two rounds, on two successive days

- In random order in both rounds

- Six consecutive spectra for each sample at each time

- Total number of scanned samples: n = 41

- Total number of stored spectra: n = 492

- Data processing with The Unscrambler 9.7 (CAMO Software AS, Oslo, Norway) and MSOffice Excel 2010 (Microsoft Co.)

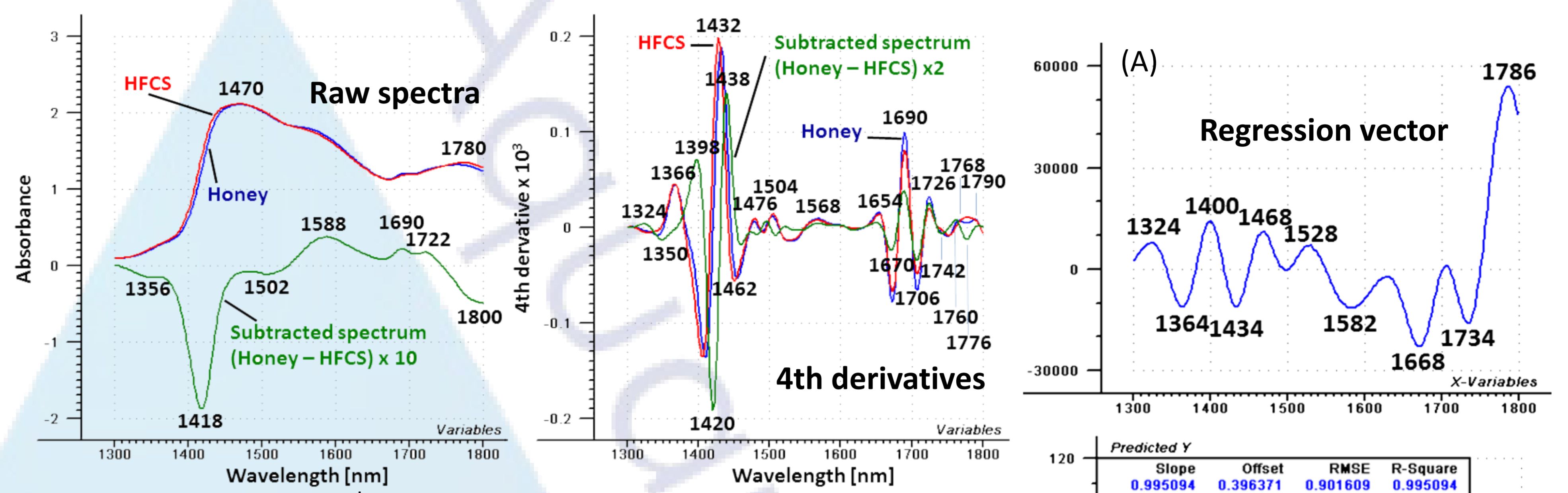
## REFERENCES

- [1] Mehryar, L., & Esmaili, M., 2011. 11th International Congress on Engineering and Food, Athens, Greece
- [2] Tsenkova, R., 2009. Journal of Near Infrared Spectroscopy, 17, 303-314.
- [3] Kinoshita et al., 2012. Scientific Reports, 2, 856; DOI:10.1038/srep00856

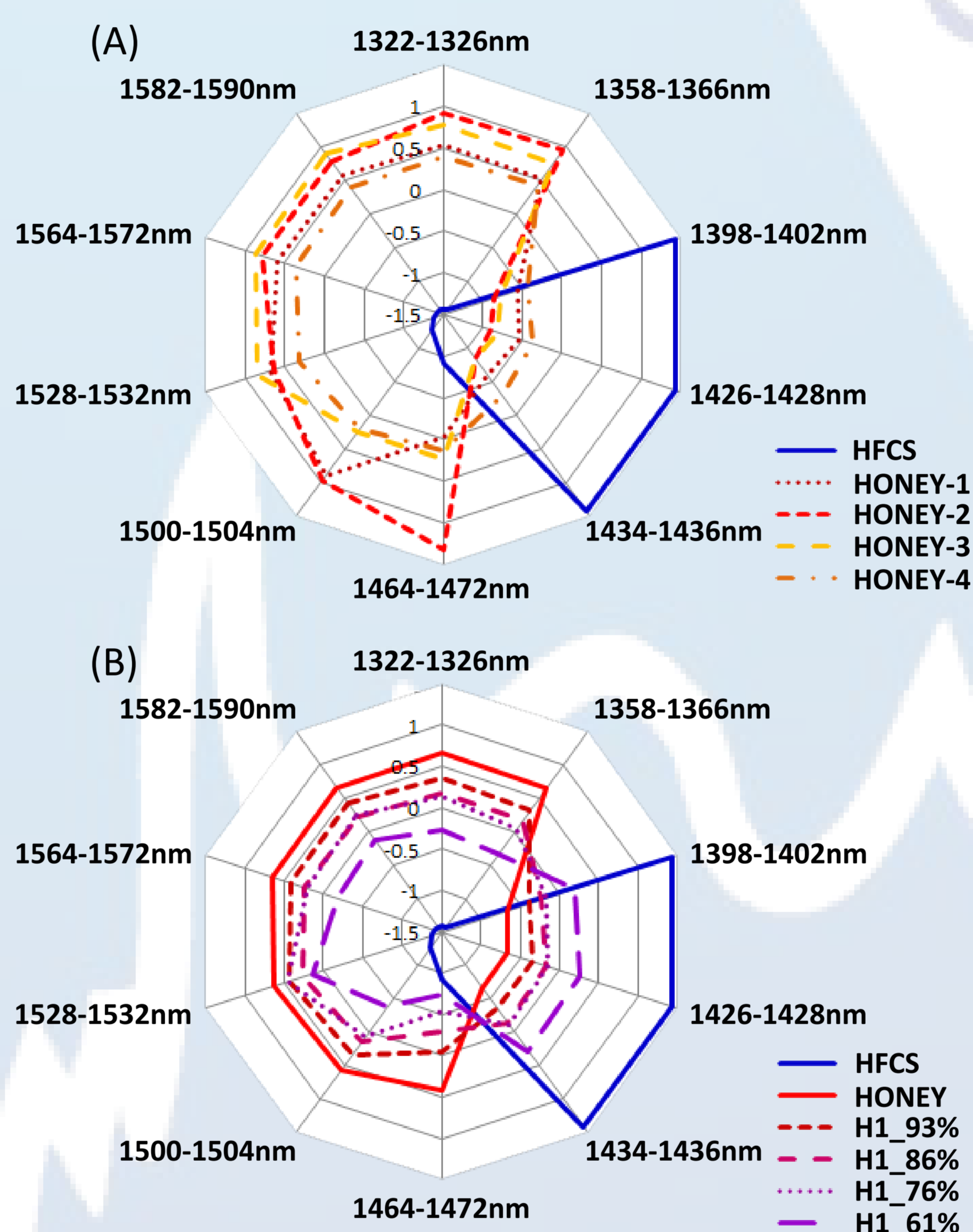
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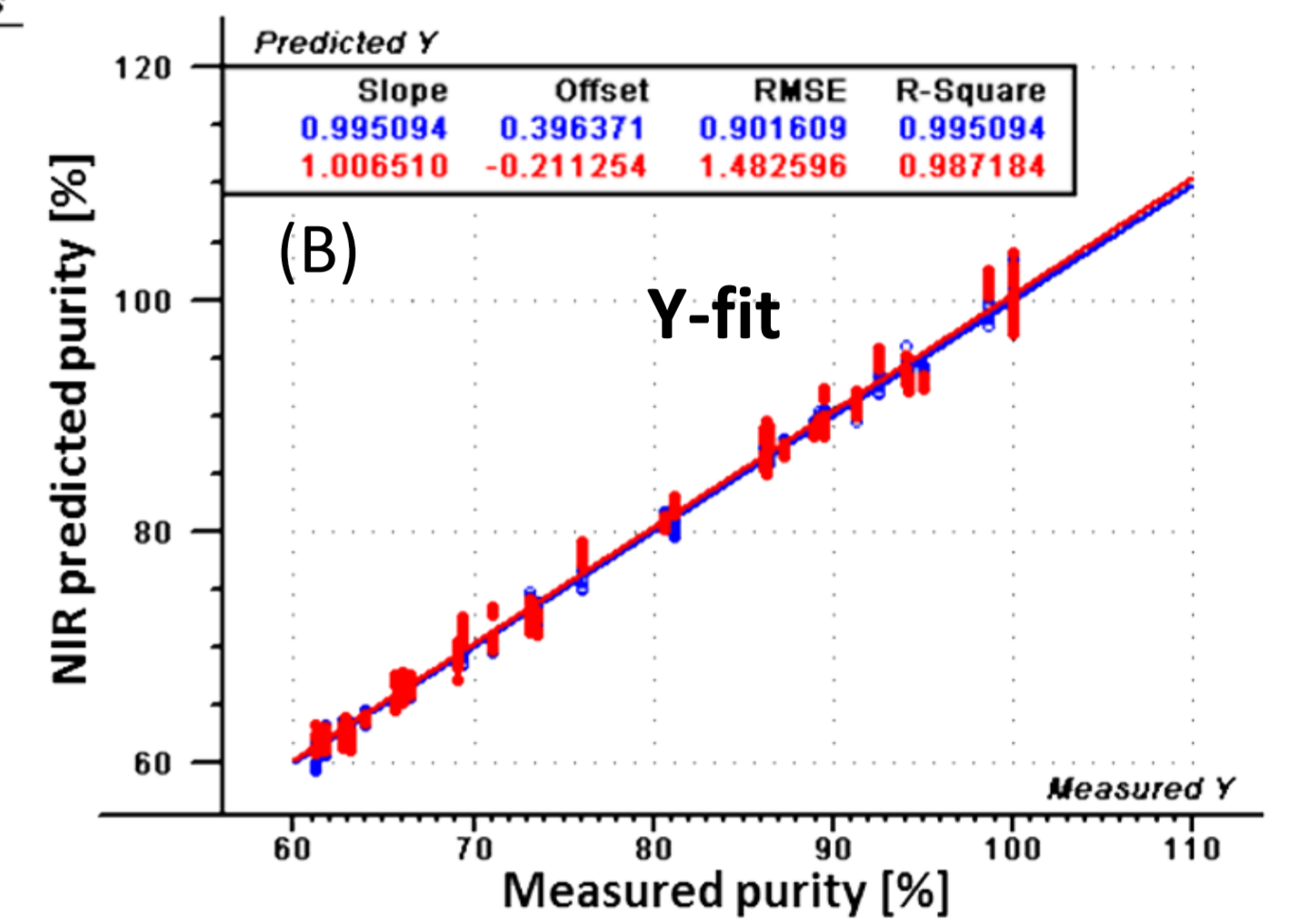
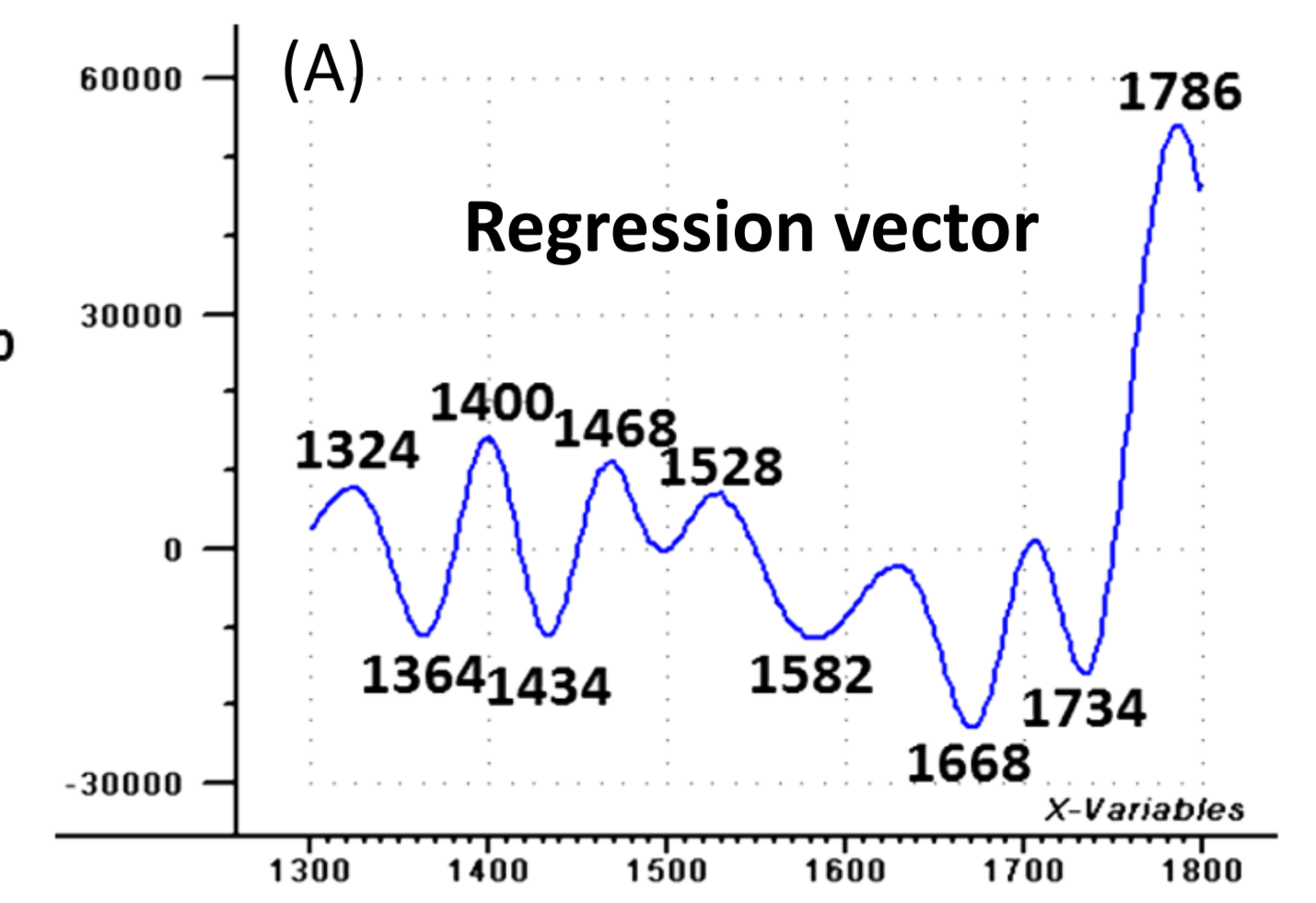
## RESULTS



Raw and 4<sup>th</sup> derivative average spectra of pure honeys and HFCS, and their subtracted spectra.



(A) Aquagram of the investigated pure *Robinia* honeys and HFCS presenting water spectral pattern. (B) Aquagram of adulterated mixtures of one honey (H1-purity%) plotted with the mean graph of the pure honeys and graph of HFCS.



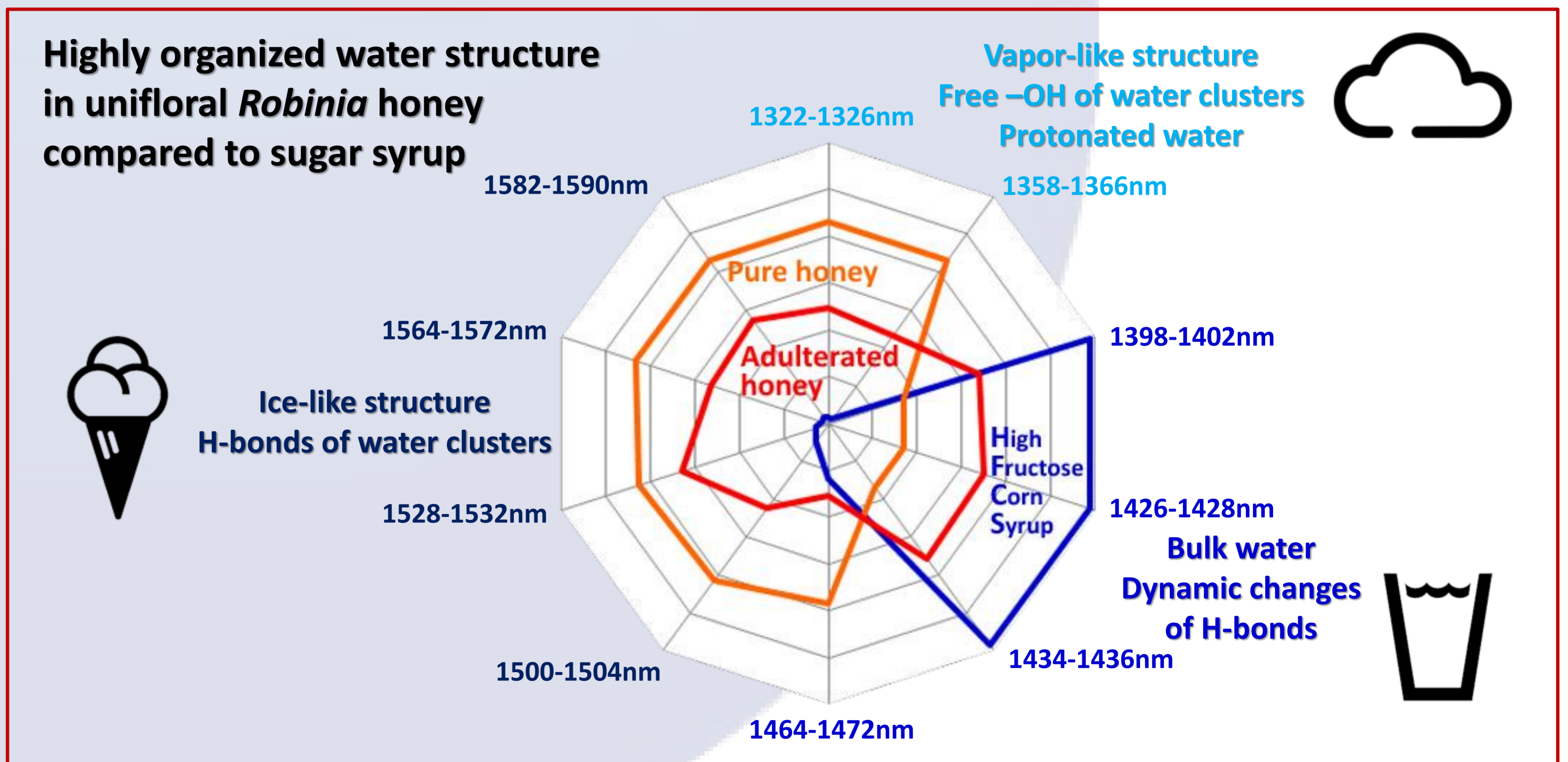
(A) Regression vectors of PLSR models with (B) the results of calibrations and cross-validations on purity (honey percentage against HFCS) of adulterated honey samples.

Characteristic changes at water matrix coordinates describe the water spectral pattern.

Visualization with aquagrams [3]:

$$A = \frac{a - \mu}{\sigma}$$

A = Aquagram value  
a = absorbance after MSC  
 $\mu$  = average of all spectra  
 $\sigma$  = SD of all spectra



## SUMMARY & CONCLUSIONS

- Quick analytical tests were developed for detecting HFCS adulteration of *Robinia* honeys.
- The most accurate NIR models for predicting adulteration level with the lowest cross-validation error (RMSE<sub>CV</sub> = 1.48%) were achieved within the whole spectral range of 1300-1800nm, containing the absorption bands of both water and carbohydrates.
- Investigated unifloral *Robinia* honeys contained larger amount of water having highly organized molecular structure, than industrial sugar syrup (HFCS).
  - larger variety of molecules dissolved in the multicomponent system of honeys
  - simpler matrix of HFCS has relatively large amount of unstructured water
- Adulteration caused gradual reduction of water trimers, molecular structures facilitating the interactions with other molecules.