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1 **Transport processes resulting from metastable boiling water under Mars**  
2 **surface conditions**  
3

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45        **Liquid water or brine could exist on the martian surface today, albeit transiently and**  
46 **in a metastable state<sup>1,2</sup>. However, the positive identification of liquid water or brine on**  
47 **Mars is hampered by our limited knowledge of how metastable liquids interact with**  
48 **sloping sediments. Here, we address this lacuna through experimental simulations of**  
49 **fluid propagation and sediment transport at martian pressure. These experiments show**  
50 **that boiling of pure water at martian pressure induces grain saltation and wholesale**  
51 **slope destabilization: a hybrid flow mechanism involving both wet and dry processes.**  
52 **This effect is decreased for metastable brines, however they can instead form channels.**  
53 **We find that seeping water and brine have a higher geomorphological impact under**  
54 **martian conditions than under terrestrial conditions. This hybrid flow mechanism could**  
55 **be responsible for martian surface changes originally interpreted as either “dry” or**  
56 **“wet” and extends the suite of processes that could be responsible for currently and**  
57 **recently active features.**

58        Determining whether liquid water or brines are presently active on the surface of Mars is  
59 of importance for understanding Mars’ hydrologic cycle, the potential for extant life, and  
60 potential resources for future explorers. Because surface pressure is frequently below the  
61 triple point, liquid water is unstable on Mars today<sup>1</sup>. Chemical and thermodynamic models<sup>1,2</sup>  
62 have, however demonstrated that under certain conditions, metastable liquid water can be  
63 transiently present, and salts in solution can depress its freezing point and reduce the  
64 evaporation rate<sup>3,4</sup>. Therefore, metastable liquid water and/or brines are viable contenders for  
65 explaining present-day active processes observed on Mars<sup>e.g.5</sup>. So far, morphological and  
66 spectral investigations have not provided unambiguous proof of liquid water flowing at the  
67 martian surface, although hydrated salt spectral signatures associated with briny flows have  
68 been detected<sup>6</sup>. Process inferences by morphological analysis are often based on analogy with  
69 terrestrial landforms formed by fluid-sediment interactions, which implicitly exclude the

70 possible effects of metastability. In order to be able to interpret martian flow-like  
71 morphologies we need to answer the following fundamental questions: (1) what is the  
72 mechanism by which metastable water/brine flows over and through a granular substrate  
73 under current martian conditions, and (2) what are the resulting spectroscopic and  
74 morphological properties that could be detectable from orbital or rover observations?

75 To respond to these questions, we have performed a set of laboratory simulation  
76 experiments with the view of investigating the specific effect of martian pressure on fluid  
77 propagation and sediment transport. The most likely source of liquid brine or water for  
78 present-day processes is the melting of seasonal frost or ground ice, and deliquescence<sup>e.g.7,8</sup>,  
79 which are thought to produce only small amounts of liquid water<sup>4,8,9,10,11</sup>. We simulate this  
80 relatively low flow rates by ice melting (propagation of ~30cm/hr consistent with calculations  
81 of Kereszturi et al.<sup>11</sup>). A 70g block of ice was placed and allowed to melt at the top of a 30°  
82 slope covered with loose fine-grained sand (Supplementary Fig. 1). Two different  
83 compositions were used for the frozen block: pure water and a eutectic MgSO<sub>4</sub> brine solution  
84 (25wt%). We first conducted experiments in a Mars Chamber facility<sup>12,13</sup> to simulate optimal  
85 martian conditions for efficient ice melting<sup>1,9,14</sup>: 6.5 or 9mbar and 293K. To understand the  
86 effect of reduced martian pressure, the experiments were then repeated at terrestrial conditions  
87 (1bar, 293K).

88 Broad similarities can be observed between the experiments conducted under terrestrial  
89 and martian pressure: for all experiments, melting of the frozen block led to the formation of a  
90 linear, darker toned, downslope-oriented flowpath (Fig. 1a-d). The dark tone is brought about  
91 by the saturation of the regolith<sup>15</sup>. Most of the liquid is transported by downslope  
92 intergranular flow, or percolation, with only occasional formation of a surface liquid film  
93 (Supplementary videos 1-6). Due to the higher viscosity of dense brines, these flows have  
94 shorter and wider flowpaths than those of pure water (Fig. 2a, Supplementary Table 1). Their

95 higher viscosity is also responsible for a higher sediment-transport capacity<sup>16</sup> and they  
96 sometimes produce a short channel in the upper part of the flowpath (Fig. 1b, d,  
97 Supplementary Table 1). In the case of the briny flows, once the flow has dried, there is  
98 crystallization of salts over the flow zone.

99 Under martian pressure, ice melting produces a metastable liquid which evaporates  
100 because it boils<sup>16</sup>. The resulting loss of water leads to a shorter final flowpath compared to the  
101 ones formed under terrestrial conditions (Fig. 2a). The metastable state of the water causes (1)  
102 the percolation to stop once the frozen block has melted, whereas under terrestrial conditions  
103 percolation continues (Fig. 2a), and (2) the propagation-rate to be reduced compared to  
104 terrestrial conditions (100cm/h and 51cm/h for terrestrial and 33cm/h and 19cm/h for martian  
105 pressure for water and brine respectively).

106 Under terrestrial pressure, the morphological impact of percolating water flows is  
107 negligible to low (Fig. 2b). The most striking difference between terrestrial and martian  
108 pressure experiments is the effect of boiling at martian pressure. Boiling is first apparent at  
109 the top of the frozen block when the melting starts (Supplementary Video 1). Liquid then  
110 begins to percolate into the sand and boiling occurs most vigorously where and when it  
111 reaches the interface between the saturated sand and the atmosphere, because the temperature  
112 gradient here is the highest (Fig. 3 T1, Supplementary Videos 1 and 2). This surface boiling  
113 causes sediment saltation – presumably particles being entrained in the vapour escaping the  
114 substrate. This process gradually constructs a millimetre-high ridge ahead of the flow front,  
115 while percolation continues (Fig. 3 T2). The slope angle of the ridge increases as it grows and  
116 when it exceeds the dynamic angle of repose ( $\sim 30\text{-}35^\circ$  on Mars and Earth<sup>17</sup>) the ridge  
117 collapses, triggering a dry granular flow (Fig. 3 T3) and creating a millimetre-high arcuate  
118 trough and remnant ridge (Fig. 3 T4, Supplementary Video 2). The flow progression at  
119 martian pressure is thus characterised by repeating, successive phases of percolation and dry

120 granular flow, creating a series of ridges and troughs along the flowpath (Fig. 1c,  
121 Supplementary Fig. 2 and Supplementary Video 5). The reduction of pressure from 9mbar to  
122 6.5mbar results in more vigorous boiling (Supplementary Video 3). In this case, despite the  
123 loss of water producing a shorter flowpath a similar volume of sediment is mobilised (Fig.  
124 2b).

125 Similar mechanisms are observed for water and briny flows at martian pressure, but the  
126 resulting morphologies differ (Fig. 1c-d). The triple point of brines has a lower temperature  
127 and pressure compared to pure water, so they are more stable under martian pressure.  
128 Consequently, although briny flows are still consistently shorter and wider than pure water  
129 flows, the difference is smaller under martian pressure. (Fig. 2a, Supplementary Table 1).  
130 However, higher stability results in less vigorous boiling, less intense saltation and a lower  
131 geomorphological impact compared to pure water (Supplementary Video 5 and 6). A briny  
132 flow produces a roughened surface comprising mm-scale arcuate ridges and troughs and  
133 sometimes the formation of a short channel (Fig. 1d). For experiments performed at 6.5mbar,  
134 the brines are less stable, and in one case we observed explosive ejection of saturated  
135 sediment associated with boiling during channel-formation, but without further examples we  
136 cannot say if this is typical behavior (Fig. 2f; Supplementary Video 4; Supplementary Fig. 2,  
137 experiment 17).

138 Flowing liquid water or brine have been hypothesised to be responsible for present-day  
139 changes observed in recurring slope lineae (RSL)<sup>7</sup> (Fig. 4a), slope streaks<sup>18</sup> (Fig. 4b), gullies<sup>19</sup>  
140 (Fig. 4c) and polar dune flows<sup>20</sup> (Fig. 4d). These features occur on steep slopes (20-30°)  
141 covered by loose sediment or sand. Due to the unstable state of water, other explanations have  
142 been proposed that invoke completely dry processes, or CO<sub>2</sub> ice sublimation<sup>21,22,23</sup>. Our  
143 experiments are not intended to replicate the morphology of such features, but they do provide

144 new insights allowing us to better assess the possible involvement of liquid water/brine in  
145 their formation and infer their formation process.

146 First, our experiments clarify the lack of spectroscopic detection of liquid water or brine  
147 associated with these active features<sup>6,15</sup>. The propagation of liquid produced by melting, at a  
148 low flow rate, occurs mainly by intergranular percolation; channel-formation only leads to the  
149 presence of free surface water for a few seconds. However, the main hydrous absorption  
150 bands are only detectable if a liquid film is present<sup>15</sup> at the spatial and spectral resolution of  
151 the martian hyperspectral imaging instruments (OMEGA and CRISM). The spectral signature  
152 of intergranular water is thus under the detection threshold of current orbital instruments.  
153 Therefore, crystallized salts along the path of a briny flow<sup>6</sup> are the only potentially  
154 identifiable spectroscopic signal of the presence of liquid water flows on the current martian  
155 surface, in addition to darkening at all wavelengths.

156 Second, the resulting morphologies observed during the experiments reveal that, due to its  
157 metastable state, a small quantity of liquid water (70g here) reaching the surface can have a  
158 disproportionate geomorphological impact (Fig. 2b). The arcuate ridges and troughs observed  
159 in our experiments are too small to distinguish with the highest spatial resolution of orbital  
160 imagers (25cm/pixel, HiRISE) and could be only detected *in-situ*. If we scale our experiments  
161 for the effect of martian gravity (see Supplementary text) we find that the ridges are 2.5 times  
162 wider. Further, in our experiments, the granular flows induced by the advancing flow-front  
163 extend over more than 20cm and can be at least twice as large as the saturated zone  
164 (Supplementary video 5). Because the ridges are 3 times more voluminous under martian  
165 gravity, the granular flows will also be 3 times bigger, possibly achieving detectable sizes. At  
166 the landscape-scale we might expect larger liquid-volumes, but from our analysis we do not  
167 expect the scale of the morphologies to change only the transported volume. On Mars, the  
168 saturated area in front of the lobe can remain “unseen” from orbit because of its smaller size

169 and/or because, after complete evaporation of the water, the lobe formed by the granular flow  
170 is the only remaining morphology. This process could be playing a role in triggering slope  
171 streaks (Fig. 4b), and smooth pale fans (Fig. 4a) and large (20 m wide) slumps<sup>24</sup> located  
172 below some active RSL. Furthermore, this activity is expected to produce the gradual or  
173 incremental growth observed for RSL<sup>7,8</sup>.

174 Our experiments also point out to fundamental differences between the way in which a  
175 stable and an unstable liquid propagates through the sediment. This demonstrates that  
176 interpretation of any current activity suspected to be water-driven on Mars, cannot be based  
177 solely on terrestrial flow morphologies. Brines are often invoked to explain recent changes on  
178 Mars, not only because they have a lower melting point, but also because water is not  
179 considered stable enough to produce small-scale, slowly propagating surface morphological  
180 changes such as RSL<sup>4</sup>. Not only have we shown that very small amounts of unstable water  
181 can produce surface changes, but by demonstrating that brines do not produce a higher  
182 morphological impact than pure water (Fig. 2b) whatever the surface pressure is.

183 This work shows that ice melting on a slope under martian pressure leads to a hybrid  
184 transport process involving both wet and dry mechanisms. To date, the interpretation of most  
185 martian surface activity has been polarised into either dry or wet processes<sup>5</sup>, and unequivocal  
186 evidence for any single process has not yet been presented. Our findings, which demonstrate  
187 the hybrid flow-mechanism of metastable fluids, extend the suite of processes that could be  
188 responsible for currently and recently active features on Mars like RSL. More generally, we  
189 have shown that only small amounts of meltwater are required to transport sediment. Even  
190 though seasonal H<sub>2</sub>O deposition is much thinner than that of CO<sub>2</sub><sup>25,26</sup>, our results demonstrate  
191 that the contribution of H<sub>2</sub>O frost should not be neglected. A combined process involving both  
192 CO<sub>2</sub> and H<sub>2</sub>O frost destabilization and melt should be considered.

193



## 194 **Methods:**

### 195 **Experimental setup.**

196 For each experiment the frozen block was placed towards the top of a 30° slope covered  
197 with loose sand (Supplementary Fig. 1) and was allowed to melt at a temperature consistent  
198 with summer martian conditions<sup>9</sup> (293K). The loose sand was placed on a plastic board,  
199 providing an impermeable layer simulating the bedrock or a layer of permafrost. Sand was  
200 adhered to the board to increase the roughness preventing basal sliding. A 30° angle was  
201 selected to fit with measured RSL<sup>7</sup>, polar dune flow<sup>20</sup> average slopes and slope streak<sup>18</sup> and  
202 gully<sup>27</sup> proximal slopes. In order to investigate the effect of sand thickness, the board was  
203 covered by either 1-2mm or 3-4mm of fine sand (200-250µm). These sand thicknesses were  
204 chosen to allow only moderate infiltration into the sand. The selected grain size is consistent  
205 with typical grain sizes found by martian rovers (52% of fine-grained soil < 250µm in Gale  
206 soil<sup>28</sup>). We used pure water and brine solution to create the frozen blocks. Both were made  
207 using 70g of pure water. For the briny block we prepared a eutectic magnesium sulfate  
208 solution of 25wt% MgSO<sub>4</sub>. MgSO<sub>4</sub> has been detected on Mars<sup>29</sup>. In solution it has an  
209 intermediate viscosity between that of water and other likely martian sulfates/chlorides<sup>16</sup> and  
210 a eutectic temperature of 270K<sup>16</sup>. A lamp was placed above the frozen block to simulate  
211 heating by solar illumination, and its intensity adjusted to maintain 293K. As melting on Mars  
212 likely occurs in the sub-surface, a thin cover of basalt powder was deposited on top of the  
213 frozen block to prevent excessive sublimation.

### 214 **Instrumentation.**

215 Experiments at martian atmospheric pressure were conducted in the low pressure Mars  
216 Chamber at the Open University (Milton Keynes, UK). In order to reproduce optimal  
217 conditions for liquid formation and stability on Mars, the majority of the experiments were  
218 performed at 9mbar<sup>14</sup>, and the others were performed at the mean martian pressure of

219 6.5mbar<sup>14</sup> (Supplementary Table 1). The pressure was actively controlled using a vacuum  
220 pump and recorded every 30s. We were able to control the pressure to within better than  
221 0.5mbar and we observed no relationship between these small fluctuations in pressure and the  
222 observed process(es). The composition of the atmosphere in the Mars chamber is terrestrial  
223 and the humidity is thus larger than martian one. This mainly changes the evaporation rate  
224 and thus, the volume of produced liquid for a similar size of ice block<sup>1</sup>. The chamber is 1m in  
225 diameter and 2m long<sup>12</sup>. Each experiment was installed at ambient pressure and temperature.  
226 In order to keep the block frozen during the 30min of depressurization, the board was placed  
227 on a cold copper plate (Supplementary Fig. 1) and the insulation only started when the  
228 selected pressure was reached. Six thermocouples were used to monitor the temperature  
229 (Supplementary Fig. 1). All experimental runs were monitored and recorded using two  
230 internal webcams and one external digital camera. The flow propagation speed was calculated  
231 by using the timestamps of the video-stills and estimating the flow propagation distance with  
232 the aid of a graduated ruler in the image frame. Once each experiment had finished,  
233 photographs were taken of the surface from different angles to produce 0.5mm/pix elevation  
234 model via the “structure from motion”<sup>30</sup> technique using Agisoft Photoscan Professional  
235 software. A 1cm moving window was used to calculate the mean elevation from the resultant  
236 Digital Elevation Model of the testbed. The volumes moved were calculated within the active  
237 area by taking the difference between this surface and the original elevation model, then  
238 summing the positive and negative values.

239 Experiments at terrestrial pressure (1bar) were conducted in the cold room at Geops  
240 laboratory (Orsay, France) with the same setup (only the copper plate was removed). The cold  
241 room was set at 0°C for the preparation step. Once the isolation step was started, the room  
242 was set at 293K. Temperature was monitored by the cold room sensor and by one  
243 thermocouple pair located within 1cm of the frozen block. Photographs were taken at regular

244 intervals using a digital camera located in front of the experiments. The flow speed was  
245 estimated by using the timestamp of each photo and a direct measurement of the flow length.  
246 The volume moved in experiment 27 was calculated by measuring the area of the “gully” on  
247 an orthophoto and estimating the thickness of sediment moved. The maximum, minimum and  
248 most-likely thicknesses of erosion/deposition were assumed to be 2mm, 1mm and 1.5mm  
249 respectively.

250

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330

### 331 **Author contributions**

332 The methodology and experimental set-up was conceived and designed by MM, SJC and JG  
333 with significant advice, help and technical support from MP, KP, AM, VC, MB, LO, FC and  
334 GJ. All data analysis was done by MM with significant feedback from SJC, JG and KP. MV  
335 and FP provided data about current water ice location and deposition. JG, SJC and SC  
336 provided physical constrains and models. All authors contributed to discussion, interpretation  
337 and writing.

338

### 339 **Additional information**

340 Supplementary information is available in the online version of the paper.

341

342 **Competing financial interests**

343 The authors declare no competing financial interests.

344

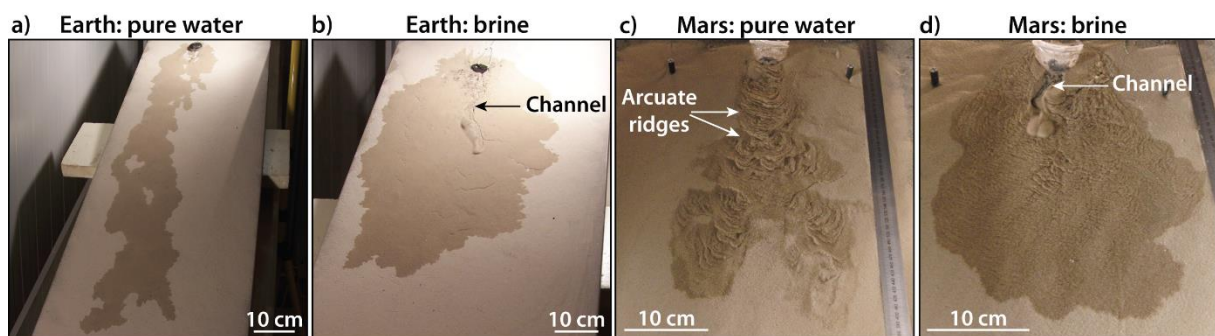
345 **Figures**

346

347 **Fig 1.** Final morphology of flows produced on a sand thickness of 1-2mm by the melting of of:

348 a) frozen water and b) frozen brine at 1bar, 293K (experiment 22 and 27, Table 1), and c)

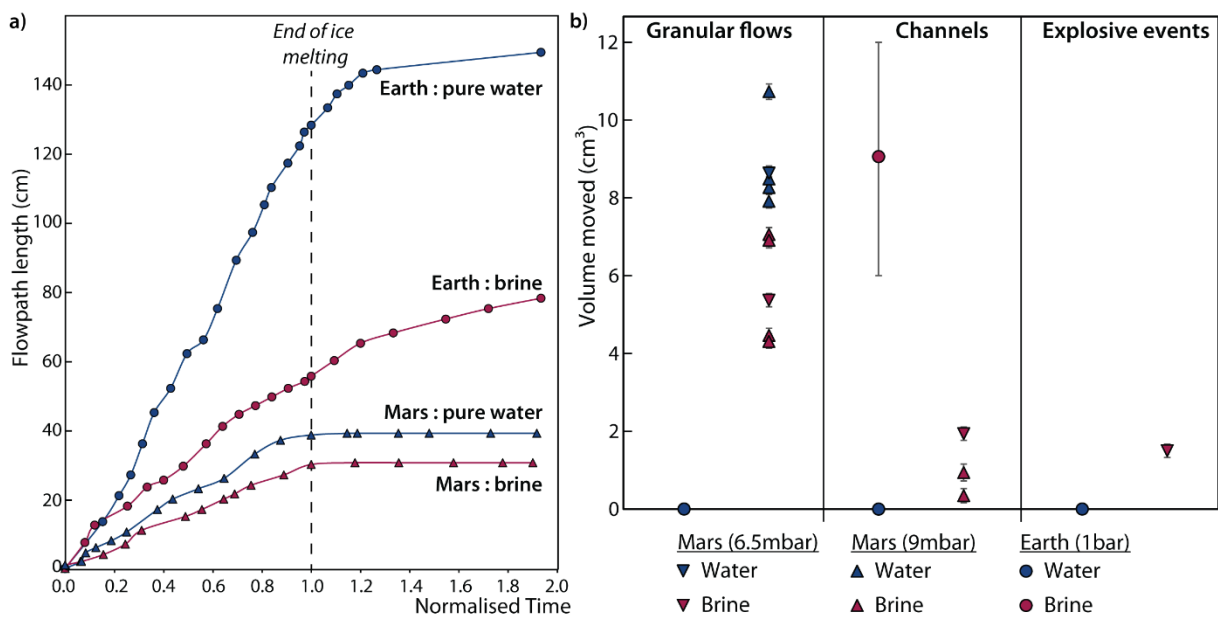
349 frozen water and d) frozen brine at 9mbar, 293K (experiment 4 and 14, Table 1).



363 **Fig 2.** Impact and evolution of the flow for different pressures and ice compositions. a)  
 364 Evolution of the flowpath length with time (time is normalised by the total duration of  
 365 melting) for same experiments as Fig. 1. b) Volume of sediment moved for all experiments  
 366 under different conditions. The error bars were calculated by taking the difference between an  
 367 interpolated reference surface and the final elevation model outside the flow and scaling-up  
 368 for the area of the flow. The error bars for brines under terrestrial conditions were calculated  
 369 by changing the thickness of erosion/deposition in the original calculation by  $\pm 0.5\text{mm}$  and re-  
 370 performing the volume estimate.

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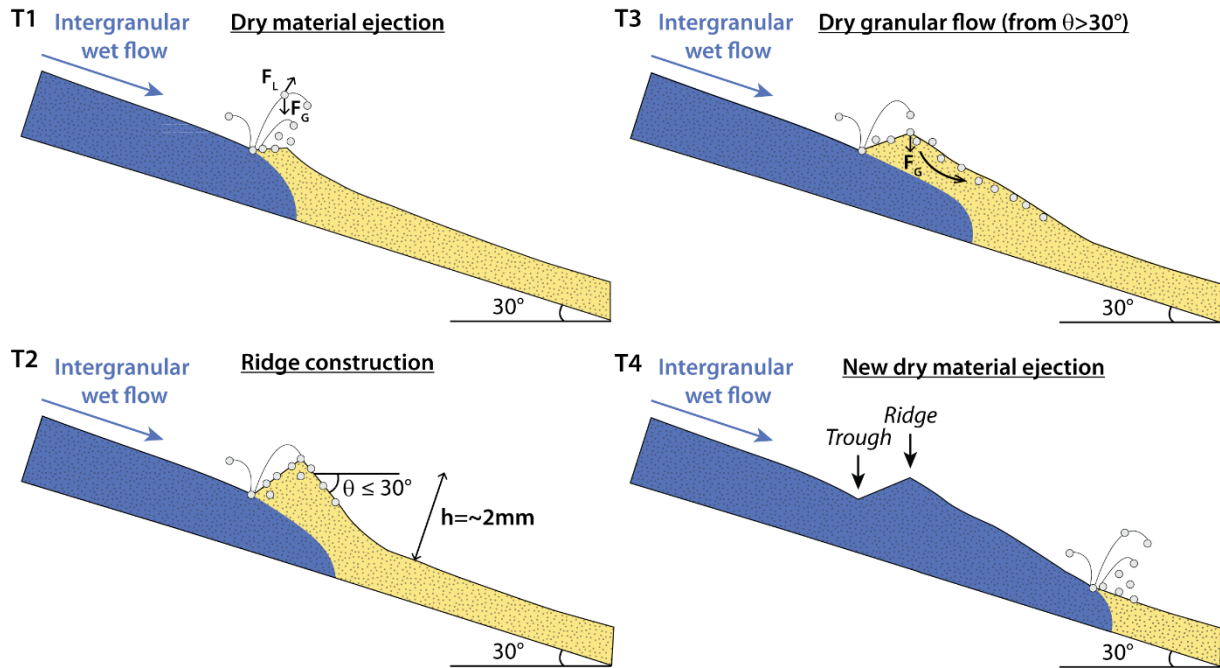
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379 **Fig 3.** Interpretative cross-sections detailing the mechanism of liquid water propagation at  
380 martian pressure. Blue areas correspond to saturated sand where water is infiltrating and  
381 yellow areas correspond to dry sand.

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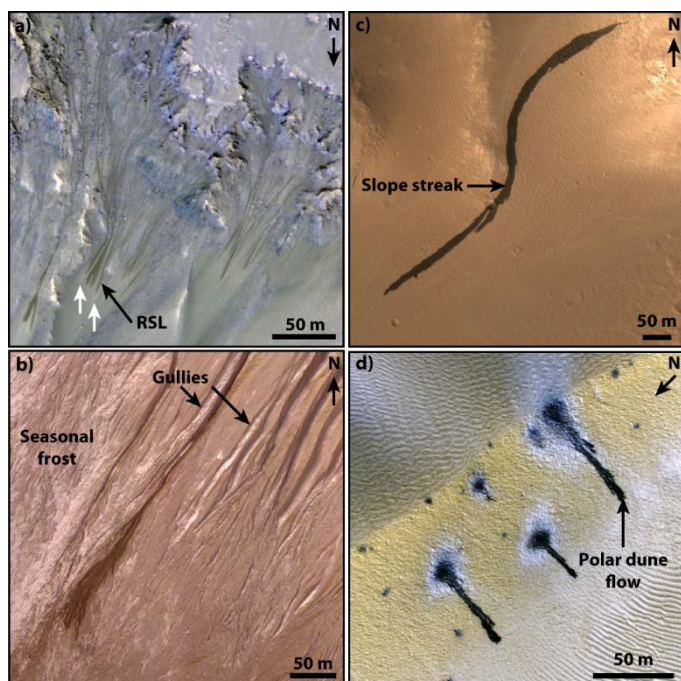
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395 **Fig 4.** Examples of current surface changes on Mars: a) Recurring Slope Lineae (RSL)  
396 (HiRISE image: ESP\_022689\_1380, center coordinates: 41.6°S, 202.3°E), the white arrows  
397 indicate smooth pale fans below the RSL, b) slope streaks (HiRISE image:  
398 ESP\_021527\_1960, center coordinates: 15.8°N, 238.2°E), c) gullies (HiRISE image:  
399 ESP\_027567\_1425, center coordinates: 37.4°S, 229°E) and d) polar dune flows (HiRISE  
400 image: PSP\_003386\_1080, center coordinates: 72°S, 179.4°E).



401

# 1 Supplementary information

2

## 3 Supplementary Discussion

### 4 (1) Saltation by boiling

5 We can split the process of grain saltation by boiling into two stages: first the initial  
6 acceleration of the sand grain induced by the boiling process, and second the ballistic  
7 trajectory for the grain which causes a downslope ridge to be built.

8 The process of boiling is a complex, but well-studied physical phenomenon. Treated simply,  
9 it is the conversion of a liquid into a gas, however it is more complex than evaporation,  
10 because it occurs at temperatures in excess of the boiling point (superheating), which causes  
11 the formation of bubbles, creating a multiphase (gas-liquid) fluid. In our experiments, the  
12 liquid water we produced via melting was already at the triple-point, therefore any  
13 temperature in excess of 273K constitutes superheat. The liquid water was in contact with  
14 material between 288K and 293K, corresponding to a superheat of 288-293K and therefore  
15 the water was in the “nucleate” boiling regime<sup>1</sup>, where gas can be released as jets or columns.  
16 In order to assess whether it is indeed these expulsions of gas which result in grain saltation,  
17 we first need to estimate the speed at which a sand grain needs to be ejected in order for it to  
18 obtain the necessary height and distance to form the ridges we observe in our experiments.

19 We therefore performed some simple calculations of the ballistic trajectories of the grains  
20 under terrestrial gravitational acceleration (Fig. S3). Similarly to Brož et al. (2014)<sup>2</sup> this finite  
21 difference model takes into account the influence of the weight of the sand grain ( $F_G = mg$ ;  
22 Fig. 3) and the drag force exerted by the martian atmosphere ( $F_D = 0.5\rho_f A C_D v^2$ ), where  $m$  is  
23 the mass of the sand grain,  $g$  the gravity,  $\rho_f$  the fluid density,  $A$  is the surface area of the grain,  
24  $C_D$  is the drag coefficient and  $v$  the velocity of the sand ejected. We used  $C_D = 1.18$  after de  
25 Blasio (2011)<sup>3</sup> and a  $\rho_f$  of  $\sim 0.01 \text{ kg}\cdot\text{m}^{-3}$  calculated for our experiments using the gas law. We  
26 used an angle of projection from the horizontal  $\beta$  and ejected the grains over an inclined plane  
27 with a slope of  $\theta=30^\circ$  (matching our experimental setup). We assumed spherical grains. Next  
28 we optimized the initial grain speed so that the model produced the same morphology as the  
29 experiment, as follows. In order to obtain grain-saltation which achieves the same height and  
30 width of the ridges observed in the experiment (height 5-10 mm, with an angle of repose at  
31  $30^\circ$ ), an initial grain speed of  $\sim 0.35 \text{ m}\cdot\text{s}^{-1}$  is required. In order to accelerate a grain to this  
32 speed, the gas velocity must be sufficient to entrain and accelerate the grain. We used the  
33 same formulation for the drag-force and constants as defined above and assumed a phase of  
34 acceleration over a distance of 1 mm (equivalent to the path-length assumed below), from this  
35 we obtained a gas speed of  $\sim 90\text{-}100 \text{ m}\cdot\text{s}^{-1}$ .

36 As stated above we believe that jets of gas produced through nucleate boiling are a logical  
37 candidate process for creating the impulsion ( $F_L$ ; Fig. 3) which accelerates the grains. Using  
38 the work of Jolly (2004)<sup>4</sup> we were able to perform some simple calculations to assess whether  
39 such a mechanism could produce gas speeds approaching the required  $\sim 90\text{-}100 \text{ m}\cdot\text{s}^{-1}$ . The  
40 work of Jolly (2004)<sup>4</sup> considers the theoretical framework of boiling in capillary tubes. In our  
41 case we consider the porosity of the sediment as the “tube”, with a path length of  $L$  through  
42 the sand layer ( $e = 1 \text{ mm}$ ). We assume its cross sectional area can be approximated by the  
43 gaps between grains, where we assume a porosity of 25% to 40%, giving an equivalent radius  
44 ( $R_{eq}$ ) of  $\sim 10 \mu\text{m}$ . We assume that the phase change occurs because the liquid experiences a  
45 decrease in pressure along the “tube”. We know that the surface of the sand is in contact with  
46 the atmosphere of the chamber ( $P_s = 600$  or  $900 \text{ Pa}$ ) whereas the vapour pressure of the liquid

47 in the sand layer is at  $T_0 = 293 \text{ K}$  ( $P_{liq} > P_{sat}(T_0) > P_s$ ). The difference between these two  
48 pressures is given by  $\delta P_v = P_{sat}(T_0) - P_s$ . If the equivalent radius is small enough we can  
49 assume to the first approximation that all the liquid is converted into vapour, when its  
50 pressure is less than, or equal to  $P_{sat}(T_0)$  along the path.

51 Therefore, using the following formulation, we can estimate the speed of the gas ( $U_{gas}$ )  
52 induced by the phase-change within the pores of the sediment:

$$53 \quad U_{gas} = (R_{eq}^2 \delta P_v) / (8 L \mu_v)$$

54 where  $\mu_v$  is the dynamic viscosity of the gas. For  $R_{eq} = 7 \times 10^{-5} \text{ m}$ ,  $\delta P_v = 1400 \text{ Pa}$ ,  
55  $L = e \sin(30^\circ)$  and for water vapour  $\mu_v = 1.8 \times 10^{-5} \text{ kg.m}^{-3}$ , we obtain a gas speed,  $U_{gas} = 91$   
56  $\text{m.s}^{-1}$ . This speed is within the range of the  $90\text{-}100 \text{ m.s}^{-1}$  required to impart an initial velocity  
57 of  $\sim 0.35 \text{ m.s}^{-1}$  to build ridges of the size observed in our experiments. Therefore we conclude  
58 that such a mechanism is physically plausible and that this kind of boiling is capable of  
59 inducing grain saltation.

60 In addition, as noted previously, in order to accelerate the grain the gas speed needs to be  
61 sufficient to overcome the weight of the grain. Therefore, this mechanism should have an  
62 upper limit on grain size, and although we have not tested this experimentally, we can further  
63 explore this limit numerically (see below).

## 64 (2) Effect of reduced martian gravity

65 The difference in gravitation acceleration between our experiments and those experienced on  
66 the surface of Mars affect a number of the physical processes present in our experiments,  
67 including: fluid propagation rates, grain ejection trajectories and finally the resulting granular  
68 flows.

69 In order to estimate the effect of martian gravity on the grain-trajectory we have built a simple  
70 numerical model. We simulate the acceleration-stage by assuming an outgoing gas-velocity of  
71  $100 \text{ m.s}^{-1}$ , as discussed above. This acceleration phase places limits on the grain size that can  
72 be ejected by this process of  $\sim 2 \text{ mm}$  under terrestrial conditions and  $\sim 4 \text{ mm}$  under martian  
73 conditions (calculated for  $\beta = 80^\circ$ ).

74 In the boiling scheme described above, capillary forces dominate over gravitational forces,  
75 therefore the outgoing gas speed should be independent of gravitational acceleration<sup>4</sup>.  
76 However, because of the reduced gravity on Mars, the counteracting weight force ( $F_G$ ) is  
77 reduced, resulting in a higher initial grain ejection speed, compared to Earth. In our  
78 simulations we have considered grain trajectories including and excluding this effect.

79 After this initial acceleration, we let the grain follow a ballistic trajectory. The results of the  
80 simulation are shown for terrestrial and martian gravity on Fig. S3, the grain travels  $\sim 2.5\text{-}3$   
81 times further under martian conditions. Because of the low atmospheric density, under both  
82 terrestrial and martian gravity, the weight is significantly higher than the drag force ( $F_G$   
83  $\gg F_D$ ), so the path followed by the grain is described by a quasi-parabolic curve, which is  
84 independent of the grain mass and size, for a given initial velocity.

85 Because the grains travel  $\sim 2.5$  times further, the resulting ridge would be at least  $\sim 2.5$  times  
86 wider. As observed in our experiments we expect this ridge to grow until it reaches the angle  
87 of repose and triggers a grain-flow, therefore the ridge would also be  $\sim 2.5$  times higher.  
88 Therefore, such a ridge would be  $\sim 3$  times more voluminous under martian gravity, resulting  
89 in ensuing granular flows with a similar increase in volume and a corresponding increase in  
90 downslope transport. As the dynamic angle of friction does not scale with gravity<sup>5,6</sup>, this

91 would mean the additional runout of such flows would be simply caused by their increased  
92 volume.

93 As discussed in more detail in other papers<sup>7,8</sup>, gravity forces dominate over capillary forces  
94 for percolation and therefore fluid percolation rates on Mars are  $\sim 1/3$  slower than on Earth  
95 due to the reduced martian gravity. The likely result of this is that ridges built by saltation  
96 could be even larger because the boiling-front does not saturate the ridge so quickly,  
97 compensating for the reduced transport from a slower-flow propagation. This compensation  
98 effect, however would need to be confirmed in the laboratory by further simulations.

### 99 **(3) Up-scaling to landscape-scale**

100 The process of grain-saltation occurs at the interface between the saturated and dry sediment,  
101 and notably at the surface. No matter how deep the sediment, or voluminous the flow, this  
102 process should still occur at the linear interface. Therefore, even at landscape-scale we expect  
103 the ridges and associated granular flows to remain at the centimetre- to decimetre-scale,  
104 respectively, therefore approaching, but still below the present observable limits of orbital  
105 data. As an example, using our coupled acceleration-ballistic model described above, using  
106 the lowest reasonable grainsize of 50  $\mu\text{m}$  (the smallest size before electrostatic forces become  
107 important) and leaving the other parameters the same, we predict ridges up to 25 cm in width  
108 ( $\sim 10$  cm high) for martian gravity, which could result in up to metre-scale granular flows.

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137 **Supplementary table:**

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139 **Table S1:** List and description of all the experiments performed in the Mars Chamber (Open  
 140 University, UK, Experiments 1-21) and in the cold room (Geops, France, Experiments 22-31).  
 141 The pressure for experiments performed in the Mars Chamber was recorded every 30 seconds  
 142 and the standard deviation from the mean pressure is given for each experiment.

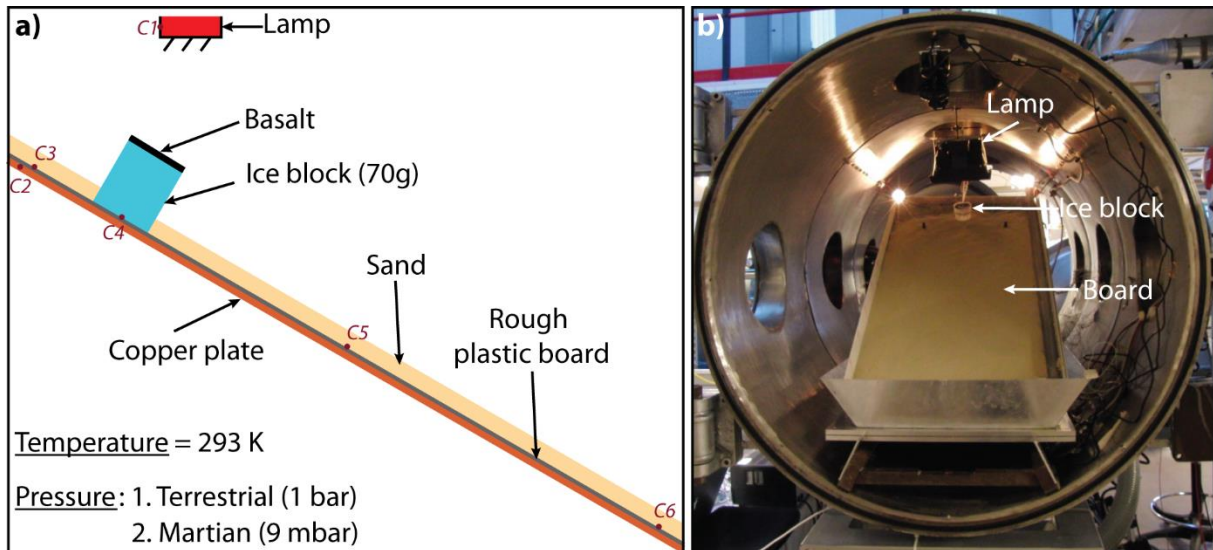
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Experiment	Pressure (mbar)	Composition	Sand thickness	Flowpath length (cm)	Flowpath maximum width (cm)	Dry granular flows	Channel
1	9.5 ± 0.53	Pure water	1-2 mm	42.5	26	Yes	No
2	9.8 ± 0.19	Pure water	1-2 mm	35.5	24	Yes	No
3	9.6 ± 0.30	Pure water	1-2 mm	44.5	27	Yes	No
4	9.6 ± 0.12	Pure water	1-2 mm	39	22.5	Yes	No
5	9.7 ± 0.22	Pure water	1-2 mm	48	23	Yes	No
6	9.8 ± 0.12	Pure water	1-2 mm	32	27	Yes	No
7	6.8 ± 0.20	Pure water	1-2 mm	25	21	Yes	No
8	29.3 ± 0.45	Pure water	1-2 mm	45	26	No	No
9	9.7 ± 0.13	Pure water	3-4 mm	32	21	Yes	No
10	9.6 ± 0.17	Pure water	3-4 mm	31	16	Yes	No
11	9.7 ± 0.10	Pure water	3-4 mm	33	16	Yes	No
12	9.7 ± 0.11	Pure water	3-4 mm	31	14	Yes	No
13	9.6 ± 0.12	Brine	1-2 mm	37	36	Yes	No
14	9.7 ± 0.17	Brine	1-2 mm	33	24	Yes	Yes
15	9.3 ± 0.41	Brine	1-2 mm	36	24	Yes	Yes
16	9.7 ± 0.16	Brine	1-2 mm	40	25	Yes	Yes
17	6.8 ± 0.21	Brine	1-2 mm	25	22	Yes	Yes
18	9.6 ± 0.09	Brine	3-4 mm	24	21.5	Yes	Yes
19	9.6 ± 0.10	Brine	3-4 mm	22	23	Yes	No
20	9.6 ± 0.12	Brine	3-4 mm	24	21	Yes	No
21	9.8 ± 0.21	Brine	3-4 mm	22	17	Yes	No
22	1000	Pure water	1-2 mm	149	20	No	No
23	1000	Pure water	1-2 mm	150	23	No	No
24	1000	Pure water	1-2 mm	101	22	No	No
25	1000	Pure water	3-4 mm	63.5	15	No	No
26	1000	Pure water	3-4 mm	67.5	21	No	No
27	1000	Brine	1-2 mm	83	35	No	Yes
28	1000	Brine	1-2 mm	83	38	No	Yes
29	1000	Brine	1-2 mm	90.5	28	No	No
30	1000	Brine	3-4 mm	67	18	No	No
31	1000	Brine	3-4 mm	73.5	22	No	No

144 **Supplementary figures:**

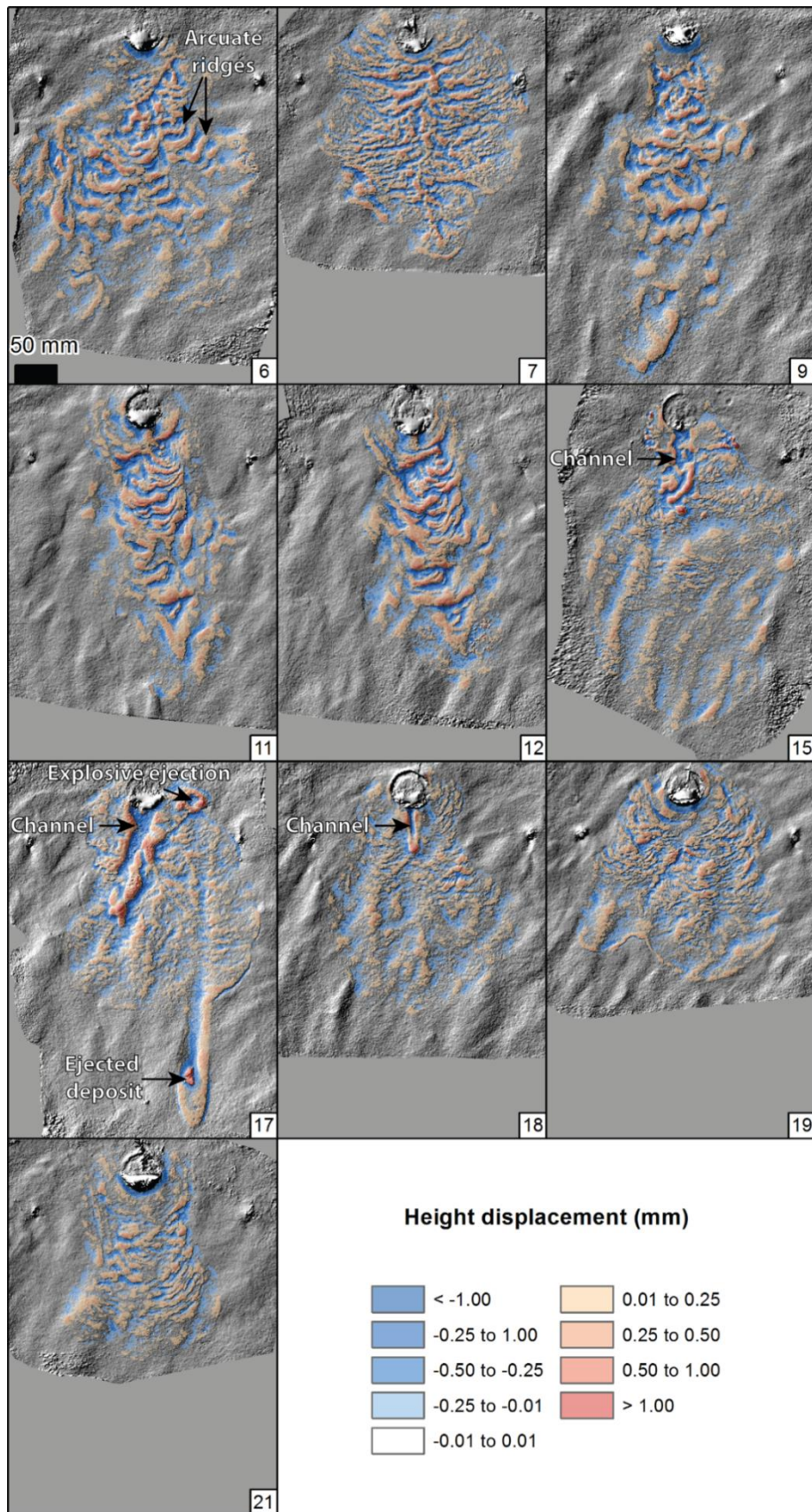
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146 **Figure S1:** The experimental setup. a) Diagram of the experimental setup. The points C  
147 represent the emplacement of all the thermocouples. b) Picture of the Martian Chamber (Open  
148 University, UK).



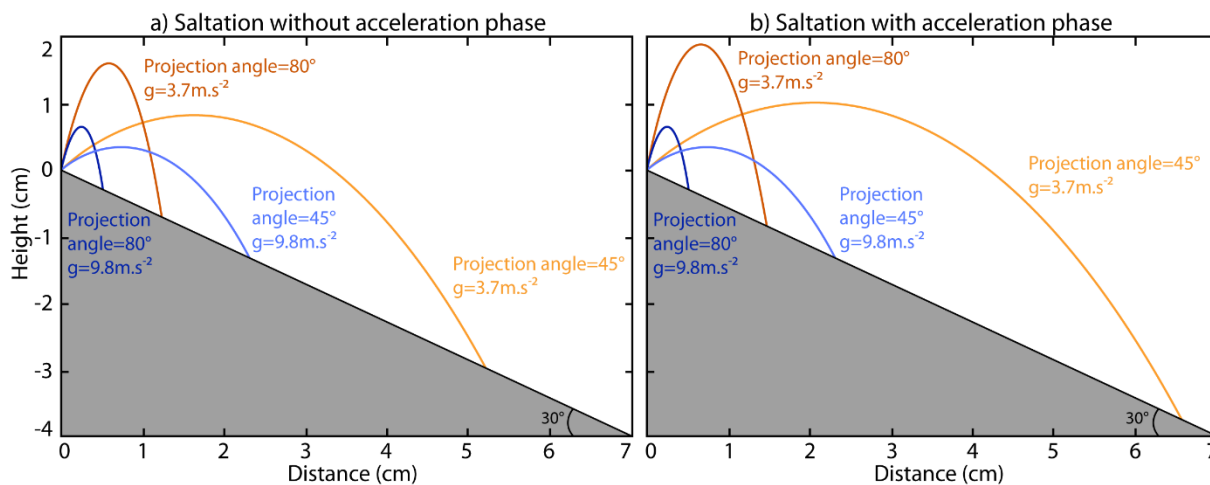
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151 **Figure S2:** Hillshaded digital elevation models with overlain difference maps (see Methods)  
 152 of the final morphologies obtained for the experiments 6, 7, 9, 11, 12, 15, 17, 18, 19 and 21  
 153 (description of the experiments on Supplementary Table 1). Colors are shown only in areas  
 154 which were used to calculate volumes in Fig. 2b.





155 **Figure S3:** Saltation trajectories for 200  $\mu\text{m}$  grains with a density of  $2600 \text{ kg.m}^3$  through an  
 156 atmosphere with a density of  $\sim 0.01 \text{ kg.m}^{-3}$  under terrestrial and martian gravitational  
 157 acceleration. a) Simulated trajectories using a constant initial speed of  $0.365 \text{ m.s}^{-1}$ . b)  
 158 Simulated trajectories using an initial speed which depends on the acceleration phase, for  
 159 Earth this results in an initial speed of  $0.365 \text{ m.s}^{-1}$ , and for Mars  $0.381 \text{ m.s}^{-1}$ .



160

161 **Supplementary videos:**

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163 **Video S1:** Movie of a flow produced by pure water ice melting at 9mbar, 293K and a sand  
164 thickness of 1-2mm (experiment 2, Table 1). Boiling water is observed at the front of the flow  
165 (associated with saltation of sand grains) and at the top of the ice block.

166 **Video S2:** Movie of a flow produced by pure water ice melting at 9mbar, 293K and a sand  
167 thickness of 1-2mm (experiment 2, Table 1). Production of dry granular flows induced by  
168 saltation of sand grains at the front of the flow.

169 **Video S3:** Movie of a flow produced by water ice melting at 6.5mbar, 293K and a sand  
170 thickness of 1-2mm (experiment 7, Table 1). Vigorous boiling is observed at the front of the  
171 flow (associated with sand grain saltation) and at the top of the ice block.

172 **Video S4:** Movie of a flow produced by briny ice melting at 6.5mbar, 293K and a sand  
173 thickness of 1-2mm (experiment 17, Table 1). A briny flow starts under the sand and lifts the  
174 sand surface. An explosive ejection of saturated sediment, which is associated with the  
175 formation of a channel, occurs when the liquid brine reaches the surface (at 10 seconds). The  
176 hole produced by a previous ejection can be seen in the upper right.

177 **Video S5:** Movie of a flow produced by pure water ice melting at 9mbar, 293 K and a sand  
178 thickness of 3-4mm (experiment 10, Table 1). Overview of the flow evolution with time  
179 showing large dry granular flows (up to 15 cm long) and the formation of a series of arcuate  
180 ridges and troughs.

181 **Video S6:** Movie of a flow produced by briny ice melting at 9mbar, 293 K and a sand  
182 thickness of 3-4mm (experiment 18, Table 1). Overview of the flow evolution with time,  
183 showing small dry granular flows (1 – 3 cm long, from 09:34:29 to 09:40:49) and formation  
184 of a channel (5 cm long, at 09:44:28).