

# Acetic acid, cyano-, methyl ester

<b>Other names:</b>	Cyanoacetic acid methyl ester Malonic methyl ester nitrile Methyl 2-cyanoacetate Methyl cyanoacetate Methyl cyanoethanoate Methylester kyseliny kyanoctove USAF kf-22
<b>Inchi:</b>	InChI=1S/C4H5NO2/c1-7-4(6)2-3-5/h2H2,1H3
<b>InchiKey:</b>	ANGDWNBGPBMQHW-UHFFFAOYSA-N
<b>Formula:</b>	C4H5NO2
<b>SMILES:</b>	COC(=O)CC#N
<b>Mol. weight [g/mol]:</b>	99.09
<b>CAS:</b>	105-34-0

## Physical Properties

Property code	Value	Unit	Source
chl	-1979.10 ± 0.46	kJ/mol	NIST Webbook
chl	-1977.00	kJ/mol	NIST Webbook
gf	-117.94	kJ/mol	Joback Method
hf	-243.30 ± 1.10	kJ/mol	NIST Webbook
hfl	-312.00	kJ/mol	NIST Webbook
hfl	-309.53 ± 0.54	kJ/mol	NIST Webbook
hfus	10.41	kJ/mol	Joback Method
hvap	66.20 ± 0.90	kJ/mol	NIST Webbook
hvap	66.23 ± 0.88	kJ/mol	NIST Webbook
hvap	56.20 ± 4.20	kJ/mol	NIST Webbook
ie	10.87 ± 0.05	eV	NIST Webbook
log10ws	-0.23		Crippen Method
logp	0.073		Crippen Method
mcvol	76.040	ml/mol	McGowan Method
pc	4000.70	kPa	Joback Method
rinpola	900.00		NIST Webbook
rinpola	900.00		NIST Webbook
tb	478.70	K	NIST Webbook
tb	473.70	K	NIST Webbook
tc	672.61	K	Joback Method
tf	260.00 ± 0.02	K	NIST Webbook

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	145.06	J/mol×K	469.29	Joback Method
cpg	150.86	J/mol×K	503.18	Joback Method
cpg	156.48	J/mol×K	537.06	Joback Method
cpg	161.89	J/mol×K	570.95	Joback Method
cpg	167.09	J/mol×K	604.84	Joback Method
cpg	172.07	J/mol×K	638.73	Joback Method
cpg	176.83	J/mol×K	672.61	Joback Method
hvapt	54.90	kJ/mol	479.00	NIST Webbook

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	388.20	K	4.80	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.52851e+01
Coeff. B	-4.35259e+03
Coeff. C	-7.06500e+01
Temperature range (K), min.	360.87
Temperature range (K), max.	507.06

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C105340&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C105340&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure: Crippen Method:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a> <a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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